

TABLE 1. Observed pK_a Values for 3,7-Diazabicyclo[3.3.1]nonanes, Determined as pK_{BH^+} for the Conjugated Acid, Spectrophotometrically in AN Solution

Bispidine	Structure	pK_a	Bispidine	Structure	pK_a
1		8.13	6		17.48
2		13.48	7		17.79
3		13.81	8		21.25
4		13.97	9		21.38
5		16.93	10		21.66

TABLE 6. Results of UV–Vis Spectrophotometric Titration Experiments in AN Solution and Assigned $pK_{a,AN}$ Values for the Conjugate Acid Forms of the Compounds 1–10

bispidine	reference base		concn, ^b 10 ⁵ M		acid ^c used	calcn method ^d	ΔpK_a^e	<i>s</i>	assigned pK_a value
	identity	pK_a^a	bispidine	reference base					
1	3-NO ₂ -aniline	7.68	1.59	4.74	T	S	0.44	0.05	8.13
	3-NO ₂ -4-F-aniline	7.67	2.56	5.67	T	S	0.42	0.04	
	2,6-(MeO) ₂ -pyridine	7.64	2.49	7.11	T	S	0.48	0.07	
	2,4-F ₂ -aniline	8.39	2.58	8.33	T	S	-0.23	0.03	
	2-Cl-pyridine	6.79	3.29	13.25	T	S	1.38	0.05	
2	2,6-Cl ₂ -4-NO ₂ -C ₆ H ₂ -P ₁ (pyrr)	14.43	1.73	3.67	M	NV/S	-0.95	0.1	13.48
	2,6-(NO ₂) ₂ -C ₆ H ₃ -P ₁ (pyrr)	14.12	1.50	5.22	M	NV/S	0.65	0.05	
3	2,4-(NO ₂) ₂ -C ₆ H ₃ -P ₁ (pyrr)	14.88	3.57	3.19	M	NV/S	-0.95	0.1	13.81
	2,6-(NO ₂) ₂ -C ₆ H ₃ -P ₁ (pyrr)	14.12	2.78	7.53	M	NV	-0.35	0.05	
	2,6-Cl ₂ -4-NO ₂ -C ₆ H ₂ -P ₁ (pyrr)	14.43	2.74	4.03	M	NV	-0.62	0.05	
4	3-NH ₂ -pyridine	14.17	1.67	6.63	M	NV/S	-0.44	0.1	13.97
	2,4-(NO ₂) ₂ -C ₆ H ₃ -P ₁ (pyrr)	14.88	2.31	3.07	M	S	-0.90	0.03	
	2,6-(NO ₂) ₂ -C ₆ H ₃ -P ₁ (pyrr)	14.12	1.91	2.91	M	S	-0.17	0.01	
	2,6-Cl ₂ -4-NO ₂ -C ₆ H ₂ -P ₁ (pyrr)	14.43	2.50	3.68	M	S	-0.47	0.01	
5	2-NO ₂ -4-CF ₃ -C ₆ H ₃ -P ₁ (pyrr)	16.54	2.45	1.78	M	NV	0.39	0.05	16.93
	2-NO ₂ -5-Cl-C ₆ H ₃ -P ₁ (pyrr)	17.27	3.00	2.08	M	NV	-0.34	0.05	
	2-NO ₂ -4-Cl-C ₆ H ₃ -P ₁ (pyrr)	17.68	2.66	2.83	M	NV	-0.76	0.05	
6	2-NO ₂ -4-CF ₃ -C ₆ H ₃ -P ₁ (pyrr)	16.53	1.98	3.61	M	NV	0.96	0.05	17.48
	2-NO ₂ -5-Cl-C ₆ H ₃ -P ₁ (pyrr)	17.27	4.16	1.85	M	NV/S	0.19	0.07	
	4-NMe ₂ -pyridine	17.95	1.43	4.50	M	NV/S	-0.47	0.1	
	4-NMe ₂ -pyridine	17.95	2.67	5.02	T	NV/S	-0.46	0.1	
	2-NO ₂ -4-Cl-C ₆ H ₃ -P ₁ (pyrr)	17.68	1.88	3.87	M	NV/S	-0.19	0.1	
7	2,5-Cl ₂ -C ₆ H ₃ -P ₁ (pyrr)	18.52	1.13	1.86	M	S	-0.70	0.04	17.79
	4-pyrrolidinylpyridine	18.33	1.04	3.23	M	NV/S	-0.59	0.1	
	4-NMe ₂ -pyridine	17.95	1.42	3.32	M	NV	-0.17	0.06	
	4-NO ₂ -C ₆ H ₄ -P ₁ (pyrr)	18.51	1.12	2.85	M	S	-0.70	0.06	
8	PhP ₁ (dma)	21.25	1.24	2.49	M	NV	0.03	0.06	21.25
	4-Br-C ₆ H ₄ -P ₁ (pyrr)	21.19	1.84	1.14	M	NV/S	0.08	0.06	
	PhP ₁ (dma)Me	21.03	1.14	2.04	M	NV/S	0.23	0.1	
	4-CF ₃ -C ₆ H ₄ -P ₁ (pyrr)	20.16	1.49	1.61	M	NV/S	1.01	0.1	
	PhTMG	20.84	1.54	2.71	M	NV	0.43	0.06	
	2-Cl-C ₆ H ₄ -P ₁ (pyrr)	20.17	2.34	2.74	M	NV/S	1.2	0.1	
9	PhP ₁ (dma)	21.25	1.79	3.62	M	NV/S	0.11	0.05	21.38
	4-Br-C ₆ H ₄ -P ₁ (pyrr)	21.19	1.79	2.99	M	NV/S	0.21	0.05	
	PhP ₁ (dma) ₂ Me	21.03	2.17	2.67	M	NV	0.36	0.05	
	PhTMG	20.84	2.05	3.40	M	NV/S	0.52	0.1	
	PhP ₁ (pyrr)	22.34	4.39	3.87	M	NV/S	-0.63	0.1	
	PhP ₁ (dma)	21.25	4.07	3.17	M	NV/S	0.41	0.05	
10	PhP ₁ (dma)	21.25	6.45	3.53	T	NV/S	0.42	0.05	21.66
	4-Br-C ₆ H ₄ -P ₁ (pyrr)	21.19	3.58	3.14	M	NV/S	0.49	0.1	
	PhP ₁ (dma) ₂ Me	21.03	4.11	2.67	M	NV/S	0.60	0.07	
	PhTMG	20.84	3.97	4.32	M	NV/S	0.77	0.07	
	PhP ₁ (dma)	21.25	6.45	3.53	T	NV/S	0.42	0.05	

^a Reference 44. ^b Concentration of bispidine and reference base in mixture. ^c Abbreviation of the acid titrated with: M = CH₃SO₃H, T = CF₃SO₃H.

^d Calculation method: NV, bispidine as “nonvisible”, ΔpK_a calculated on molar basis; S, calculated from UV–vis spectra. ^e $\Delta pK_a = pK_a(\text{bispidine}) - pK_a(\text{reference base})$.